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The Structure of Pu<sub>5</sub>Ru<sub>3</sub>\*

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**Abstract.** Pu<sub>5</sub>Ru<sub>3</sub>, *I4/mcm*, *Z*=4, *a*=10.745 (3), *c*=5.719 (2) Å,  $\rho_c=15.07$  g cm<sup>-3</sup>. This structure is of the *D8<sub>m</sub>*, W<sub>5</sub>Si<sub>3</sub> type.

**Introduction.** The Pu–Ru phase diagram given by Ellinger, Land & Gschneidner (1967) shows five intermetallic compounds; Pu<sub>19</sub>Ru, Pu<sub>3</sub>Ru, Pu<sub>5</sub>Ru<sub>3</sub>, PuRu and PuRu<sub>2</sub>. The structures of the first two compounds are unknown. PuRu has the CsCl structure and PuRu<sub>2</sub> the MgCu<sub>2</sub> structure. It is the purpose of this note to report the structure of Pu<sub>5</sub>Ru<sub>3</sub>.

An alloy containing 39 at.% Ru was prepared by arc melting followed by a heat treatment for 10 days at 800°C. The alloy was crushed and many of the fragments were found to be single crystals. Preliminary precession photographs showed the crystals to be tetragonal, space group *I4/mcm* and probably of the *D8<sub>m</sub>* or W<sub>5</sub>Si<sub>3</sub> structure type. Reflections *hkl* were present only for *h+k+l=2n* and *0kl* only for *l=2n*. Lattice constants [*a*=10.745 (3), *c*=5.719 (2) Å, Mo *Kα*<sub>1</sub>=0.70926 Å] and intensities were measured with graphite-monochromated Mo *Kα* radiation on a Picker automatic diffractometer. The details of data collection have been described elsewhere (e.g. Cromer & Larson, 1972). Reflections were measured for *h, k* > 0, all values of *l* and for  $2\theta \leq 60^\circ$ . Equivalent general reflections were thus measured eight times. The shape of the irregular crystal fragment was approximated by a six-sided polygon. Absorption corrections ( $\mu=500$  cm<sup>-1</sup>) were applied by the Busing & Levy (1957) method using a modification of Burnham's (1962) program. The crystal volume was  $4.2 \times 10^{-4}$  mm<sup>3</sup> and transmission factors varied from 0.039 to 0.183. An index defined as  $R_F = \sum_n \sum_i |\bar{F}_n - F_{l,n}| / \sum_n \bar{F}_n$  was 0.087 where  $F_n = \sum w_i F_i / \sum w_i$  and the summations are over the *i* equivalent measurements with weight *w<sub>i</sub>* of the reflection *F<sub>n</sub>*. This

rather poor agreement between equivalent reflections is due to the difficulty of defining the shape of a rough, irregular crystal fragment. *F<sub>i</sub>* was considered observed if  $(I - B) \geq 2\sigma(I) = 2[I + B + (0.014I)^2]^{1/2}$ . Of the 287 unique reflections measured, 265 were observed.

Because the compound was thought to belong to a known structure type, full-matrix least-squares refinement was begun immediately. Relativistic Hartree–Fock scattering factors computed by Cromer & Waber (1974) were used along with the anomalous dispersion values of Cromer & Liberman (1970). The assumption of the structure type was shown to be correct and the final positional and thermal parameters are given in Table 1. For these parameters,  $R = \sum |\Delta F| / \sum F_o = 0.066$  and  $R_w = [\sum w(\Delta F)^2 / \sum w F_o^2]^{1/2} = 0.073$ , with unobserved reflections omitted.\*

Table 2. Interatomic distances in Pu<sub>5</sub>Ru<sub>3</sub> (Å)

Pu(1)–2Pu(1)	2.860	Ru(1)–2Ru(1)	2.860
–8Pu(2)	3.453 (1)	–8Pu(2)	2.899 (1)
–4Ru(2)	2.815 (4)		
Pu(2)–2Pu(1)	3.453 (1)	Ru(2)–2Ru(2)	3.966 (7)
–Pu(2)	3.000 (2)	–2Pu(1)	2.815 (4)
–2Pu(2)	3.371 (1)	–2Pu(2)	2.839 (4)
–2Pu(2)	3.532 (2)	–2Pu(2)	2.913 (2)
–2Pu(2)	3.566 (2)	–4Pu(2)	3.247 (1)
–2Ru(1)	2.899 (1)		
–Ru(2)	2.839 (4)		
–Ru(2)	2.913 (2)		
–2Ru(2)	3.247 (1)		

**Discussion.** The interatomic distances are listed in Table 2 and the unit-cell contents and coordination polyhedra are shown in Fig. 1. If the midpoint between

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Table 1. Final least-squares parameters for Pu<sub>5</sub>Ru<sub>3</sub>

The anisotropic temperature factor is  $\exp[-(h^2\beta_{11} + k^2\beta_{22} + l^2\beta_{33} + hk\beta_{12} + hl\beta_{13} + kl\beta_{23})]$ .  $\beta$  values are  $\times 10^5$ .

	<i>x</i>	<i>y</i>	<i>z</i>	$\beta_{11}$	$\beta_{22}$	$\beta_{33}$	$\beta_{12}$
Pu(1)	0	$\frac{1}{2}$	$\frac{1}{4}$	255 (16)	$\beta_{11}$	3432 (140)	
Pu(2)	0.0831 (1)	0.2195 (1)	0	201 (13)	161 (13)	1189 (44)	12 (15)
Ru(1)	0	0	$\frac{1}{4}$	220 (29)	$\beta_{11}$	938 (161)	
Ru(2)	0.1596 (3)	$\frac{1}{2} + x$	0	322 (25)	$\beta_{11}$	1902 (144)	82 (58)

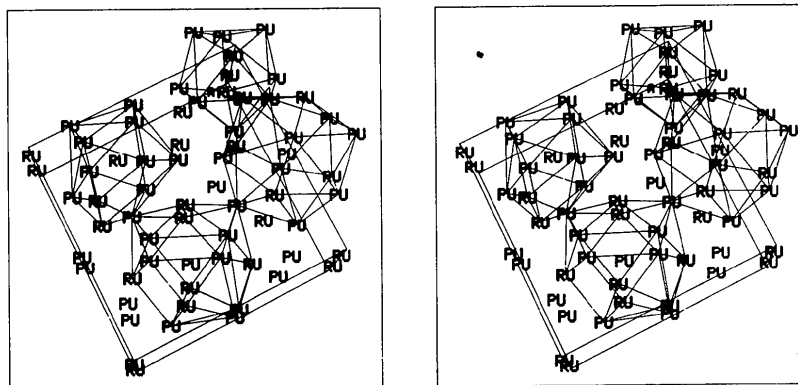


Fig. 1. View of the unit-cell contents of  $\text{Pu}_5\text{Ru}_3$  and the four polyhedra. The origin is at the upper rear corner where the positive  $a$ ,  $b$ , and  $c$  directions are labeled. The upper polyhedron is that of  $\text{Ru}(1)$  at  $0.6596, 0.1596, \frac{1}{2}$ ; the right polyhedron is  $\text{Pu}(1)$  at  $0, \frac{1}{2}, \frac{1}{4}$  and the lower polyhedron is  $\text{Pu}(2)$  at  $0.5831, 0.7195, \frac{1}{4}$ .

two atoms is closer to these atoms than to any other atoms these atoms are defined as neighbors. The polyhedra shown in Fig. 1 are the convex polyhedra formed by the neighbors of the central atom.

Other Pu intermetallic compounds known to have this structure type are  $\text{Pu}_5\text{Si}_3$  (Cromer, Larson & Roof, 1964) and the ternary phase  $(\text{Pu}, \text{Ce})_5\text{Co}_3$  (Larson, Roof & Cromer, 1964). This latter compound has an  $a$  axis of  $10.730 \text{ \AA}$ , virtually the same as that of the present compound, but a rather shorter  $c$  axis of  $5.383 \text{ \AA}$ . This short  $c$  axis crowds the large cerium atoms in the  $0, \frac{1}{2}, \frac{1}{4}$  position and produces some disorder and vacancies. This disorder was manifested by an anomalously high apparent thermal motion along the  $c$  axis.

virtually isotropic for its ten neighbors are nearly equidistant.  $\text{Pu}(1)$  has two close  $\text{Pu}(1)$  neighbors at  $2.859 \text{ \AA}$  directly above and below along the  $c$  axis. In spite of this proximity,  $B_3$ , parallel to  $c$ , is much larger than the other axes, and a disorder similar to that found in  $(\text{Pu}, \text{Ce})_5\text{Co}_3$  was suspected.

A further refinement with an occupancy parameter added for  $\text{Pu}(1)$  suggested that its site was only 92% occupied.  $R$  was reduced to 0.0633 and  $R_w$  to 0.0679. This result should be accepted with caution and perhaps represents an 'over refinement', because the agreement index is substantially smaller than the index obtained from comparison of equivalent reflections.

All calculations were made with a CDC 7600 computer using the LASL system of crystallographic programs. We are indebted to V. O. Struebing for preparation of the alloy.

Table 3. *Thermal ellipsoids in  $\text{Pu}_5\text{Ru}_3$*

	$B_i$	Angles ( $^\circ$ ) relative to direct cell axes		
		$\alpha$	$\beta$	$\gamma$
Pu(1)	1.1 (1) $\text{\AA}^2$	0	90	90
	1.1 (1)	90	0	90
	4.9 (2)	90	90	0
Pu(2)	0.9 (1)	6.5	83.5	90
	0.7 (1)	96.5	6.5	90
	1.6 (1)	90	90	0
Ru(1)	1.1 (1)	0	90	90
	1.1 (1)	90	0	90
	1.2 (2)	90	90	0
Ru(2)	1.7 (2)	45	45	90
	1.3 (2)	135	45	90
	2.5 (2)	90	90	0

The major axes of the thermal ellipsoids of  $\text{Pu}_5\text{Ru}_3$  and their orientations are given in Table 3. For  $\text{Pu}(2)$  and  $\text{Ru}(2)$ ,  $B_3$  is the largest axis. These atoms have no neighbors directly above and below in the  $c$  direction, but  $\text{Pu}(2)$  has five neighbors and  $\text{Ru}(2)$  has four neighbors in the plane normal to  $c$ . The motion parallel to  $c$  is therefore less restricted. The motion of  $\text{Ru}(1)$  is

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